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## Synthesis of 2,3-benzoxepins by sequential cyclopropanation/ring-enlargement reactions of benzopyrylium triflates with diazoesters

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**Abstract**—Functionalized 2,3-benzoxepins were prepared by cyclopropanation of benzopyrylium triflates with diazoesters and subsequent TFA-mediated ring enlargement.

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Oxepins and benzoxepins are of pharmacological relevance, since they play an important role in the biosynthesis and metabolism of mono- and polycyclic aromatic compounds. For example, oxepin  $\bf A$  has been isolated from cultures of the wood-rotting fungus *Phellinus tremulae*. The 2,3-benzoxepin  $\bf B$  represents a mammalian metabolite of dibenz[a,j]anthracene.

A number of syntheses of oxepins have been reported, which include ring-enlargement reactions of pyran derivatives.<sup>1</sup> For example, 3-hexen-6-olides have been prepared by cyclopropanation of silyl ketene acetals of 5-pentanolides, to give oxabicyclo[4.1.0]heptanes, and subsequent ring enlargement.<sup>4,5</sup> The reaction of pyrylium salts with lithiated diazoesters has been reported to give 4-diazomethyl-4*H*-pyrans, which were transformed into oxepins by Pd(0)-catalyzed ring enlargement.<sup>6</sup> Very recently, Reddy et al. have reported the

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synthesis of benzoazepines by reaction of diazoesters with activated quinolines and isoquinolines.<sup>7</sup> Herein, we wish to report a convenient synthesis of functionalized 2,3-benzoxepins by cyclopropanation of benzopyrylium triflates<sup>8,9</sup> with diazoesters and subsequent TFA mediated ring enlargement.

Our first attempts to induce a cyclopropanation of chromone (1a) by copper(II) triflate catalyzed reaction with diazoester 2 failed. 10,11 We have found that the desired cyclopropane 3a could be prepared in up to 71% yield by copper(II) triflate catalyzed reaction of 2 with benzopyrylium triflate A, which was generated in situ by reaction of **1a** with Me<sub>3</sub>SiOTf<sup>8,12</sup> (Scheme 1). The presence of the copper salt was mandatory. Compound 3a was obtained as a single diastereomer; the relative configuration was established by analysis of the coupling constants. Treatment of 3a with trifluoroacetic acid (TFA) resulted in formation of the 2,3-benzoxepin 4a via intermediates **B** and **C**.<sup>13</sup> 2,3-Benzoxepins, such as **4a**, generally reside in their oxepin form; 1 rearrangement of the latter to the corresponding naphthalene oxide would interrupt the aromaticity of the annulated benzene moiety.

The preparative scope of our methodology was studied (Scheme 2, Table 1). The reaction of **2** with methylsubstituted chromones **1b** and **1c** afforded the cyclopropanes **3b** and **3c**, which were transformed into the 2,3-benzoxepins **4b** and **4c**, respectively. The 2,3-benzoxepins **4d**—**f** were prepared from the bromo-, chloro- and

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Scheme 1. Synthesis of 2,3-benzoxepin 4a.

$$R^2$$
 $R^3$ 
 $R^3$ 

**Scheme 2.** Synthesis of benzoxepins **4a**–**g**. Reagents and conditions: (i) Me<sub>3</sub>SiOTf, Cu(OTf)<sub>2</sub> (5 mol %), CH<sub>2</sub>Cl<sub>2</sub>; (ii) TFA, CH<sub>2</sub>Cl<sub>2</sub>.

Table 1. Products and yields

3, 4	$\mathbb{R}^1$	$\mathbb{R}^2$	$\mathbb{R}^3$	% (3) <sup>a</sup>	% <b>(4)</b> <sup>a</sup>
a	Н	Н	Н	71	56
b	Me	Н	Н	76	32 (56)
c	Н	Me	Н	69	27 (57)
d	Н	Br	Н	64	43 (54)
e	Н	Cl	Н	68	23 (32)
f	Н	Ph	Н	89	30 (65)
g	H	Н	Me	64	41

<sup>&</sup>lt;sup>a</sup> Yields of isolated products (in brackets: yields based on recovered starting material).

phenyl-substituted chromones 1d-f. The reaction of 2 with 3-methylchromone (1g) gave the cyclopropane 3g, which was transformed into the 2,3-benzoxepin 4g.

Scheme 3. Synthesis of benzoxepin 7. Reagents and conditions: (i)  $Me_3SiOTf$ ,  $Cu(OTf)_2$  (5 mol %),  $CH_2Cl_2$ ; (ii) TFA,  $CH_2Cl_2$ .

HO

$$CO_2Et$$
 $Ar$ 
 $Ar$ 
 $CO_2Et$ 
 $ArB(OH)_2$ 
 $CO_2Et$ 
 $CO_2Et$ 
 $CO_2Et$ 
 $CO_2Et$ 
 $CO_2Et$ 
 $CO_2Et$ 

**Scheme 4.** Synthesis of benzoxepins **4a–g**. Reagents and conditions: (i) Tf<sub>2</sub>O, pyridine, -78 to -10 °C; (ii) Pd(PPh<sub>3</sub>)<sub>4</sub> (3 mol %), K<sub>3</sub>PO<sub>4</sub> (1.5 equiv), dioxane, reflux.

**9b** (Ar = 4-(MeO)C<sub>6</sub>H<sub>4</sub>): 72%

The reaction of **2** with chromone **5**, prepared from 1-acetylnaphth-2-ol, afforded the cyclopropane **6** which was subsequently transformed into the benzoxepin **7** (Scheme 3).

The functionalization of the ring by Suzuki reactions was next studied. Benzoxepin **4a** was transformed into the triflate **8**. The Suzuki reaction of the latter with *p*-tolyl- and (*p*-methoxyphenyl)boronic acid afforded the benzoxepins **9a** and **9b** in good yields, respectively (Scheme 4).

In summary, we have reported the synthesis of functionalized 2,3-benzoxepins by cyclopropanation of benzopyrylium triflates with a diazoester and subsequent TFA mediated ring enlargement.

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- 12. General procedure: To chromone 1 (2.0 mmol) and copper(II) triflate (0.1 mmol, 5 mol %) was added Me<sub>3</sub>SiOTf (2.6 mmol) by syringe and the mixture was stirred for 30 min at 20 °C under inert atmosphere. To the mixture was added CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and, subsequently, a CH<sub>2</sub>Cl<sub>2</sub> solution (5 mL) of 2 was slowly added at 0 °C. After stirring for 2–5 h (TLC control) the solution was poured into water, the organic and the aqueous layer were separated and the latter was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent of the filtrate was removed in vacuo. The

- residue was purified by chromatography (silica gel) to give a colourless oil. Starting with **1a** (292 mg, 2.0 mmol), Me<sub>3</sub>SiOTf (578 mg, 2.6 mmol) and **2** (274 mg, 2.4 mmol), **3a** was obtained as a yellowish oil (327 mg, 71%). 

  <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz):  $\delta$  = 1.27 (t,  ${}^{3}J$  = 7.2 Hz, 3H, CH<sub>3</sub>), 2.39 (dd,  ${}^{3}J$  = 3.0 Hz,  ${}^{3}J$  = 4.8 Hz, 1H, CH), 2.80 (dd,  ${}^{3}J$  = 4.8 Hz,  ${}^{3}J$  = 6.9 Hz, 1H, CH), 4.19 (q,  ${}^{3}J$  = 7.2 Hz, 2H, CH<sub>2</sub>), 4.82 (dd,  ${}^{3}J$  = 3.0 Hz,  ${}^{3}J$  = 6.9 Hz, 1H, CH), 6.97 (dd,  ${}^{3}J$  = 8.0 Hz,  ${}^{4}J$  = 1.0 Hz, 1H, Ar), 7.08 (ddd,  ${}^{3}J$  = 7.0 Hz,  ${}^{3}J$  = 8.0 Hz,  ${}^{4}J$  = 1.0 Hz, 1H, Ar), 7.51 (ddd,  ${}^{3}J$  = 7.0 Hz,  ${}^{3}J$  = 8.0 Hz,  ${}^{4}J$  = 2.0 Hz, 1H, Ar), 7.88 (dd,  ${}^{3}J$  = 8.0 Hz,  ${}^{4}J$  = 2.0 Hz, 1H, Ar), 7.88 (dd,  ${}^{3}J$  = 8.0 Hz,  ${}^{4}J$  = 2.0 Hz, 1H, Ar), 13C NMR (CDCl<sub>3</sub>, 75 MHz):  $\delta$ <sub>C</sub> = 13.99 (CH<sub>3</sub>), 21.69 (CH), 32.92, 61.30 (CH), 61.60 (CH<sub>2</sub>), 117.68 (CH, Ar), 118.42 (C, Ar), 122.36, 126.86, 135.99 (CH, Ar), 156.75 (C, Ar), 169.85, 185.94 (CO). IR (KBr):  $\nu$  = 1726 (s), 1682 (s), 1646 (w), 1610 (m), 1467 (m) cm<sup>-1</sup>. UV (MeCN):  $\lambda$  (log  $\varepsilon$ ) = 216.19 (4.3) nm. MS (EI, 70 eV): mlz = 232 (M<sup>+</sup>, 12), 187 (4), 175 (3), 121 (12), 28 (100).
- 13. General procedure: To a CH<sub>2</sub>Cl<sub>2</sub> solution (10 mL) of 3 (1.4 mmol) was added TFA (1 mL) and a drop of water. The solution was stirred until the reaction was complete (TLC control). The solution was poured into an aqueous solution of hydrochloric acid (10%), the organic and the aqueous layer were separated and the latter was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent of the filtrate was removed in vacuo. The residue was purified by chromatography (silica gel) to give a colourless solid,  $mp = 173 \, ^{\circ}C$ . Starting with 3a (327 mg, 1.4 mmol), CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and TFA (1 mL), **4a** was isolated as a colourless solid (183 mg, 56%). <sup>1</sup>H NMR (CO(CD<sub>3</sub>)<sub>2</sub>, Colorliess solid (163 liig, 3676). If NMK (CO(CD<sub>3</sub>)<sub>2</sub>, 300 MHz)  $\delta = 1.33$  (t,  ${}^{3}J = 6.9$  Hz, 3H, CH<sub>3</sub>), 4.30 (q,  ${}^{3}J = 6.9$  Hz, 2H, CH<sub>2</sub>), 6.96 (ddd,  ${}^{3}J = 7.8$  Hz,  ${}^{3}J = 8.1$  Hz,  ${}^{4}J = 0.9$  Hz, 1H, Ar), 7.03 (dd,  ${}^{3}J = 8.4$  Hz,  ${}^{4}J = 0.9$  Hz, 1H, Ar), 7.19 (ddd,  ${}^{3}J = 8.4$  Hz,  ${}^{3}J = 8.1$  Hz,  ${}^{4}J = 1.2$  Hz, 1H, Ar), 7.29 (d,  ${}^{4}J = 0.8$  Hz, 1H, Ar), 7.77 (dd,  ${}^{3}J = 7.8$  Hz,  ${}^{4}J = 1.8$  Hz, 1H, Ar), 8.21 (d,  ${}^{4}J = 0.8$  Hz, 1H, Ar), 9.26 (s, 1H, OH). <sup>13</sup>C NMR (CO(CD<sub>3</sub>)<sub>2</sub>, 75 MHz):  $\delta_C$  = 14.63 (CH<sub>3</sub>), 60.85 (CH<sub>2</sub>), 109.47 (CH), 116.99 (CH, Ar), 117.89 (C, Ar), 120.73 (CH, Ar), 122.27 (C), 126.44, 129.85 (CH, Ar), 146.86 (OCH), 152.89 (COH), 154.42 (COAr), 163.49 (COOEt). IR (KBr):  $\tilde{v} = 3248$  (s), 3162 (w), 2981 (w), 1684 (s), 1611 (w) cm<sup>-1</sup>. UV (MeCN):  $\lambda$  (log  $\varepsilon$ ) = 262.5 (4.2), 208.2 (4.5) nm. MS (EI, 70 eV): m/z = 232 (M<sup>+</sup>, 27), 204 (5), 187 (4), 121 (8), 87 (9), 28 (100). HRMS (FT-ICR, API-ES): calcd for  $C_{13}H_{13}O_4$  ([M+H]<sup>+</sup>): 233.08139; found: 233.08249.